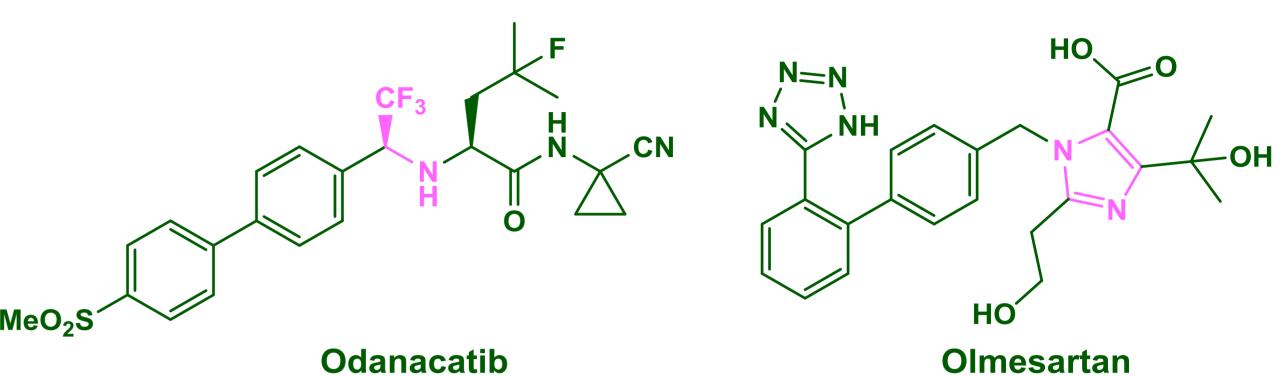


Highly substituted, trifluoromethylated imidazoles as potential scaffolds in medicinal chemistry

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Introduction

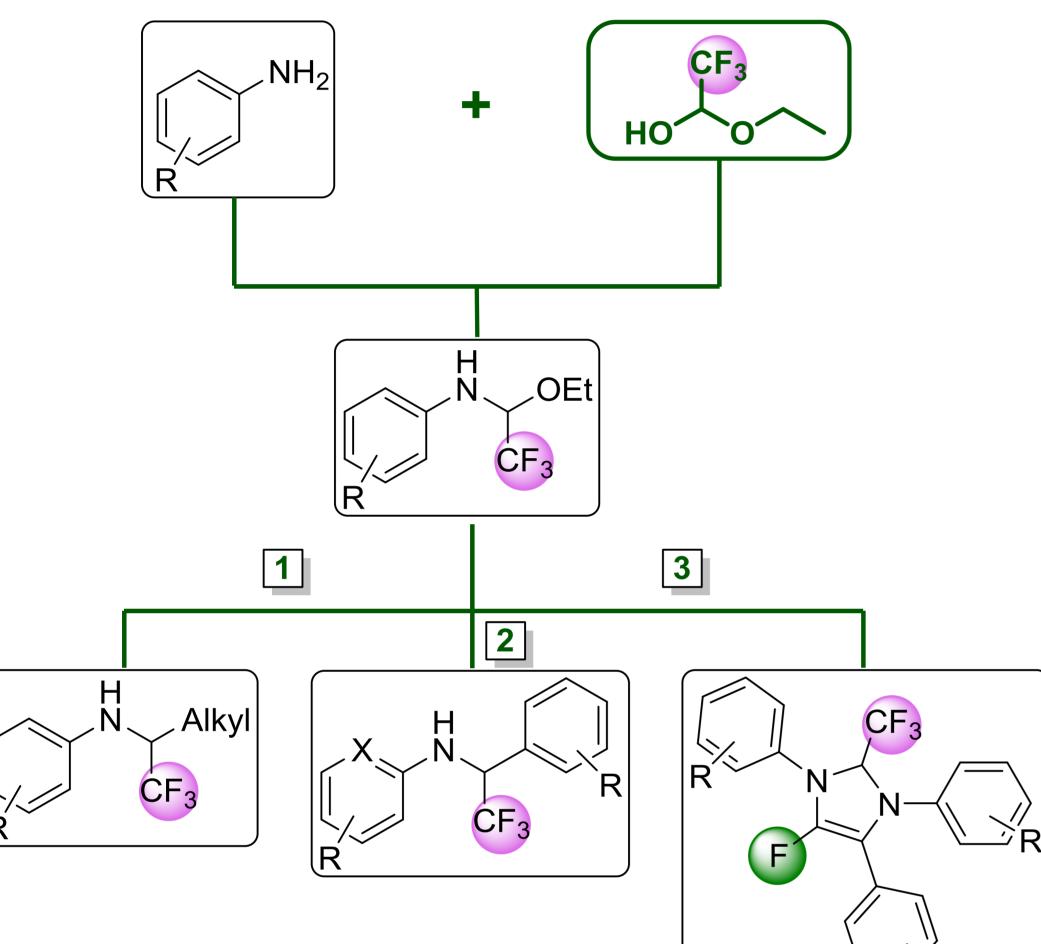
Imidazole and its derivatives have attracted considerable interest for their versatile properties in medicinal chemistry research. Next to the optimization of solubility and bioavailability parameters in active lead structures, imidazole derivatives themselves posses a broad spectrum of biological activities including anticancer and antibacterial activity.^[1] Due to the excellent pharmacological profile of fluorinated drugs, we are interested in the strategic incorporation of fluorine in trifluoromethylated amine and imidazole structures for future application in medicinal chemistry.

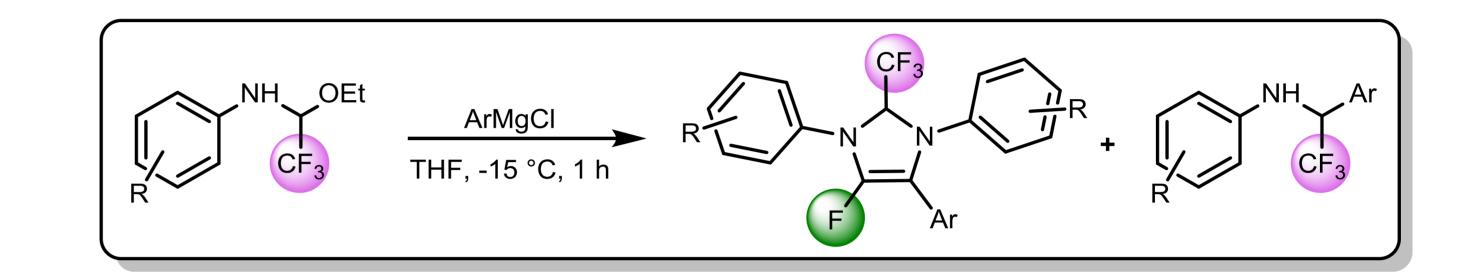


J. Y. Gauthier et al., Bioorg. Med. Chem. Lett. 2008, 18, 923.

US5616599 A1, 1997.

General Method





Trifluoromethylated imidazoles and trifluoroethyl amines are accessible from hemiaminal ethers, which are synthesized starting from the corresponding amines and trifluoroacetaldehyde ethyl hemiacetal (TFAE). The resulting hemiaminal ethers can then be converted into trifluoromethylated imidazoles or trifluoroethyl amines^[2] via addition of Grignard reagents depending on the substrates and the Grignard reagents.

1 Conversion of any hemiaminal ether with alkyl magnesium reagents

2 Conversion of electron rich and neutral substrates or hemiaminal ethers with heteroatoms in orthoposition with aryl magnesium reagents



3 Conversion of electron poor hemiaminal ethers with aryl magnesium reagents

Cl

Proposed Mechanism

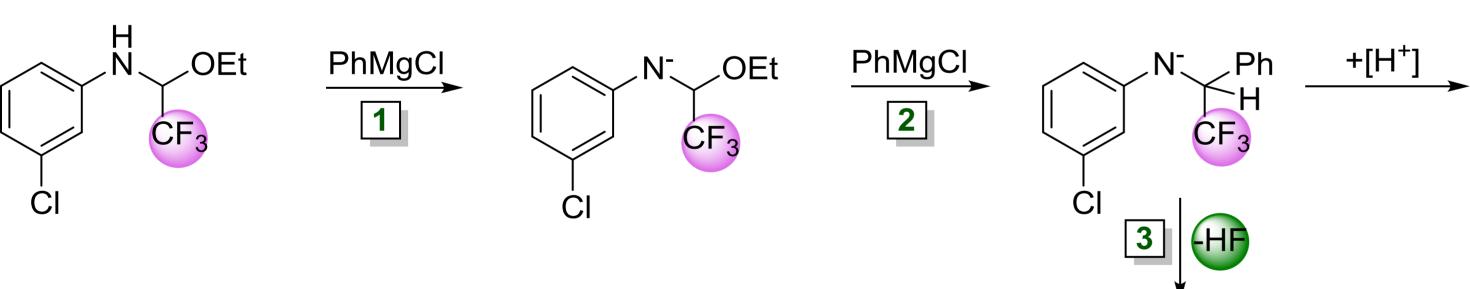


Ph

Ť-H CF₃

16%

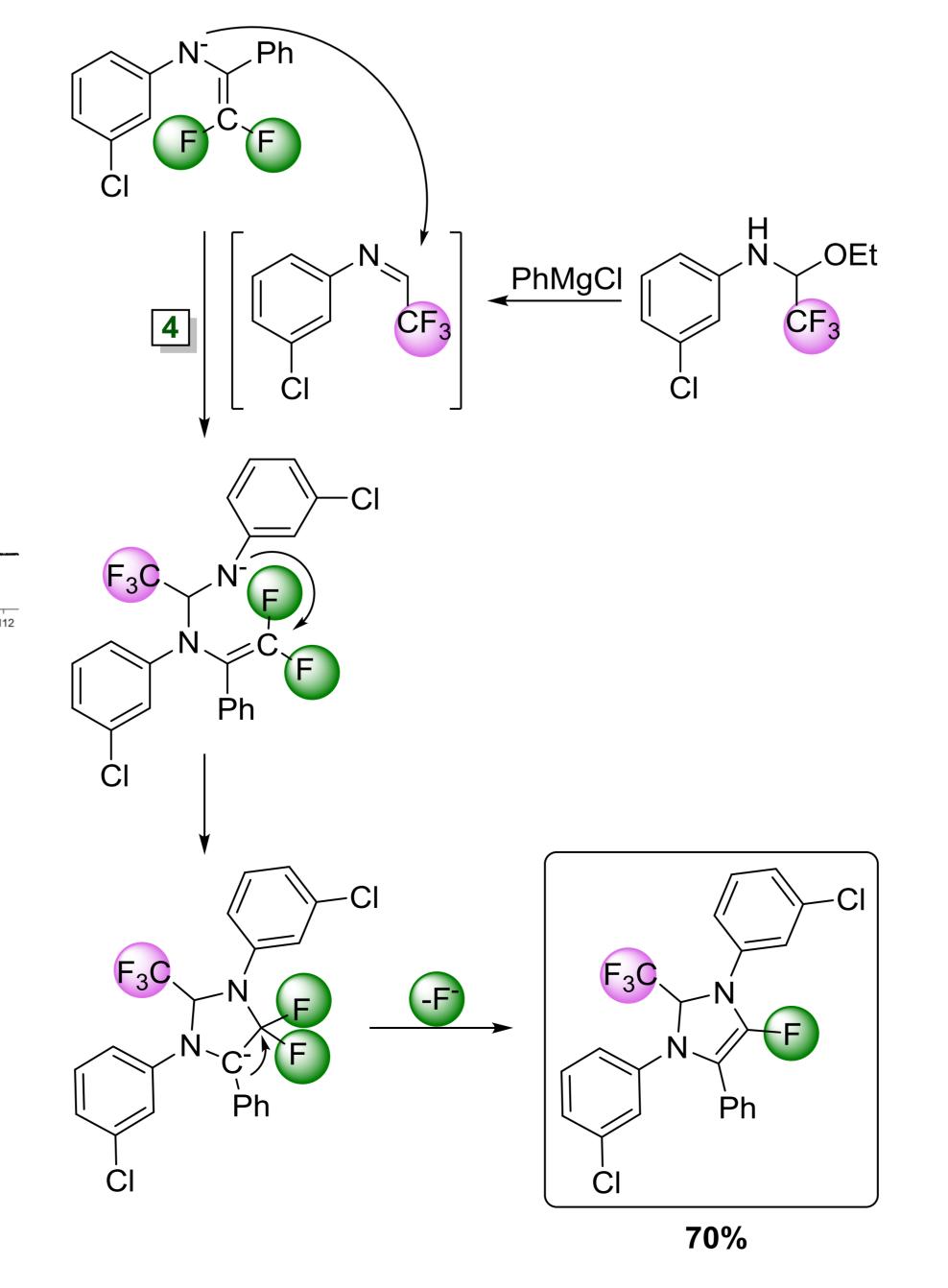
Synthesized Structures

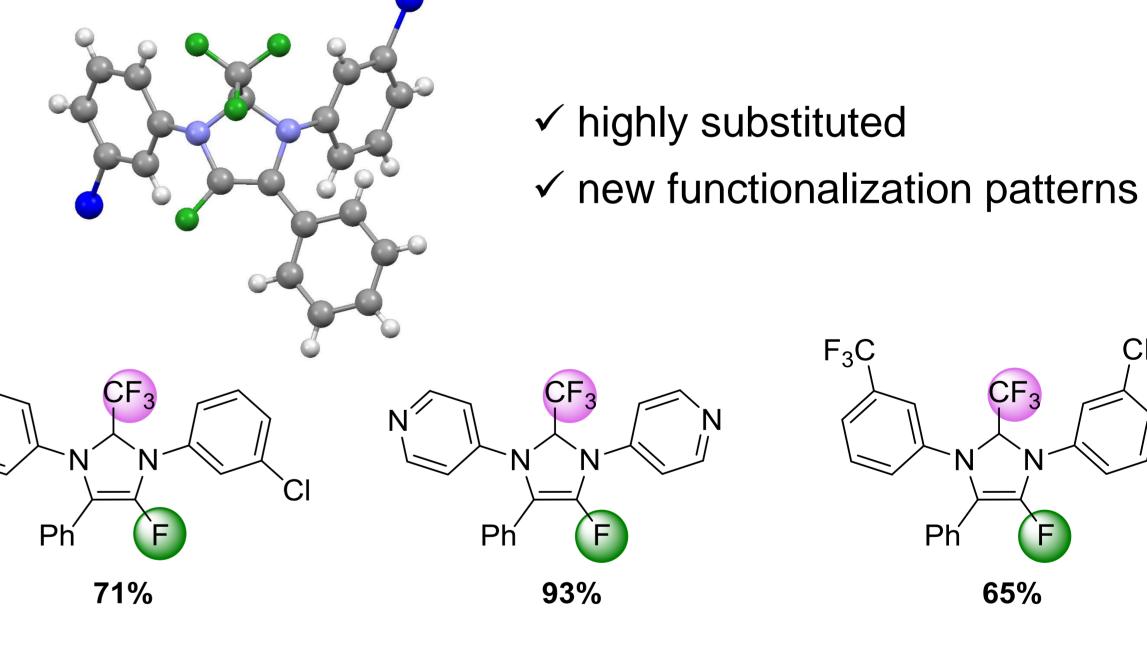


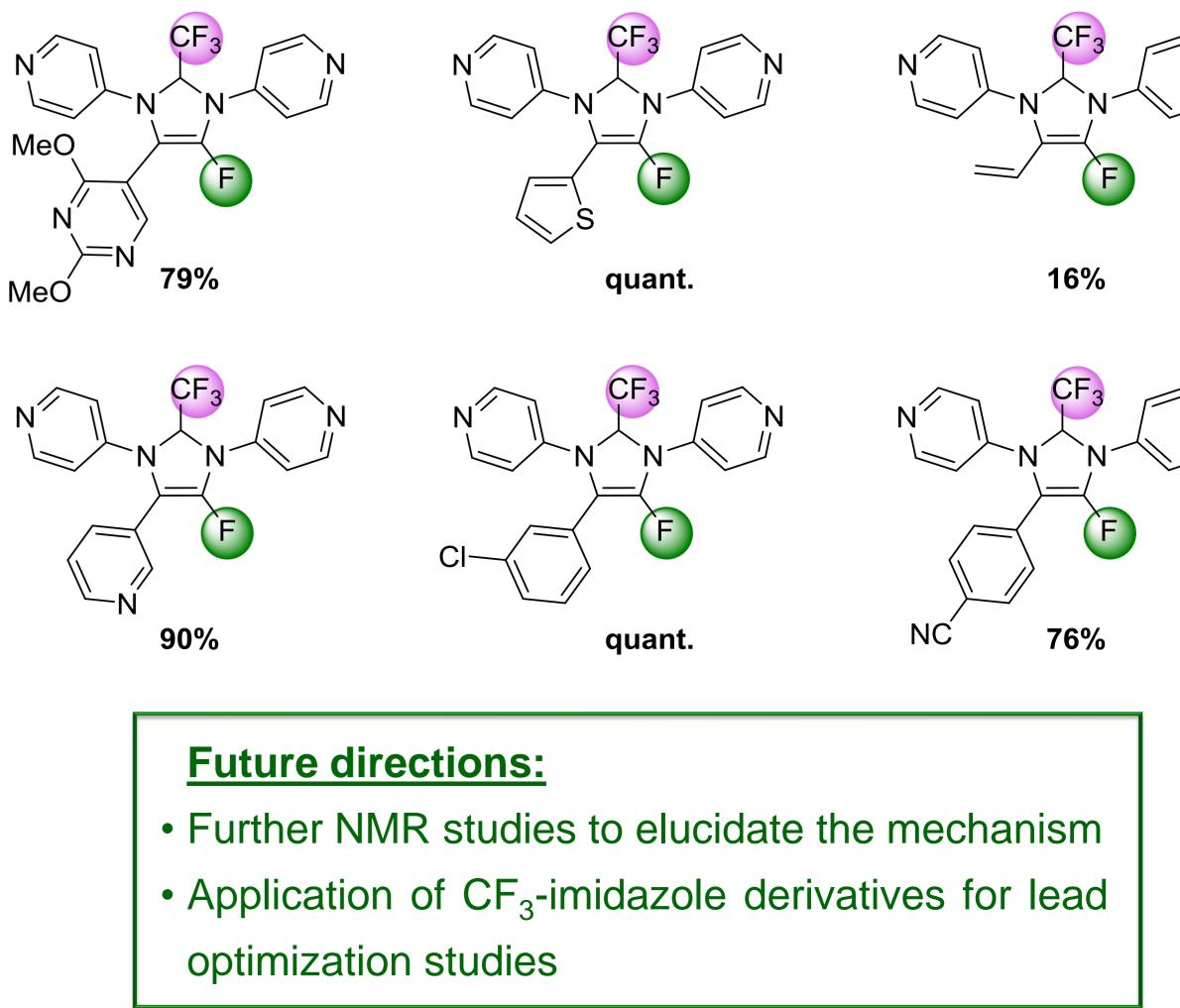
1 + 2 PhMgCl acts as base and as nucleophile HF NMR-3 elimination assumed after experiments at low temperatures:^[3]

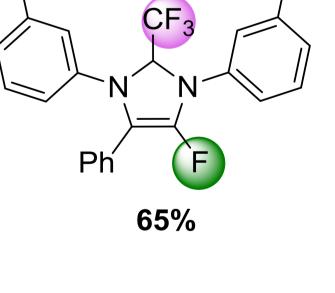
¹⁹F-NMR:



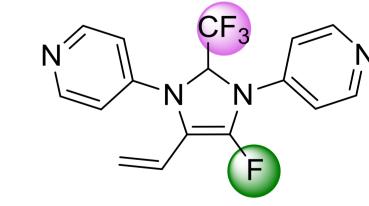




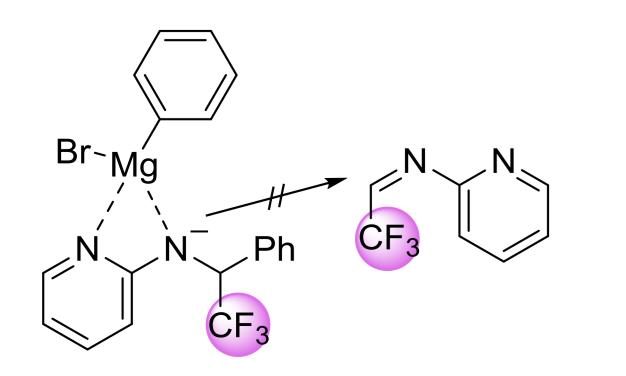




 CF_3



-70 -72 -74 -76 -78 -80 -82 -84 -86 -88 -102 -104 -106 -108 -110 -112 4 Hemiaminal ethers with heteroatoms in ortho-position hindered by are complexation with Mg:^[4]



References: [1] Zhang et al., Med. Chem. Rev. 2014, 34, 2, 340. [2] Deutsch et al., RSC Advances 2014, 4, 9288–9291. [3] Kirij et al., J. Fluorine Chem. 2008, 129, 14. [4] Ortu et al., Inorg. Chem. 2013, 52 (21), 12429.